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REVIEW ANSWER KEY

1. State the relationship between force, work/energy, and power.

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Factors . The Mole

Concept. 7. a) 13.6 mol

As b) 0.0732 mol Ne c)

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1.12 mol Li d) 3.49×10^{-3} mol S
9. a) 35.4 mol He
b) 0.0606 mol Ga
c) 1.000 mol Mg
d) 1.73 mol Cu

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Terms in this set (41)

The charge on an ion

is. Either positive or

negative. According to

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the octet rule, a calcium atom has a tendency to. Lose two electrons.

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molar mass of C_6H_6O
= 94.11 g 21. a. molar
mass of SO_3 = 80.07 g
49.2 mg SO_3 1 g 1000
mg 1 mol 80.07 g =
6.14 10^{-4} mol SO_3 b.
molar mass of PbO_2 =

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239.2 g 7.44×10^4 kg
PbO 2 1000 g 1 kg 1
mol 239.2 g = 3.11
105 mol PbO 2 c. molar
mass of CHCl_3 =
119.37 g 59.1 g CHCl_3
1 mol 119.37 g =
0.495 mol CHCl_3 d.
molar mass of $\text{C}_2\text{H}_3\text{Cl}_3$
= 133.39 g 3.27 ...

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would like to use, you
could also to open it
and start customizing it
immediately! You will
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a premium account
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